Sample Question Paper 2023-24

Sample Question Paper 2023-24 Class X Science (Subject Code – 086)

Max. Marks: 80 Time Allowed: 3 hours

General Instructions:

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. Section A consists of 20 objective type questions carrying 1 mark each.
- iv. Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- v. Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- vi. Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

Section-A

Select and write the most appropriate option out of the four options given for each of the questions 1 - 20. There is no negative mark for incorrect response.

| | | of the questions 1 - 20. There is no negative mark for incorrect response. | | | | |
|---|--------------------|--|-------|--|--|--|
| | Q. Nos. | Questions | Marks | | | |
| Ó | Q. Nos. | Test tube containing solution of barium chloride | 1 | | | |
| | | Identify the product which represents the solid state in the above reaction. | | | | |
| | a) Barium chloride | | | | | |
| | | b) Barium sulphate | | | | |
| | | c) Sodium chloride | | | | |
| | d) Sodium sulphate | | | | | |
| | 2 | The colour of the solution observed after 30 minutes of placing zinc metal to copper sulphate solution is a) Blue b) Colourless c) Dirty green d) Reddish Brown | 1 | | | |
| | 3 | Mild non-corrosive basic salt is a) Ca (OH) ₂ b) NaCl c) NaOH d) NaHCO ₃ | 3.IA | | | |
| | $Z_{C_{f_{i}}}$ | d) NaHCO ₃ | | | | |

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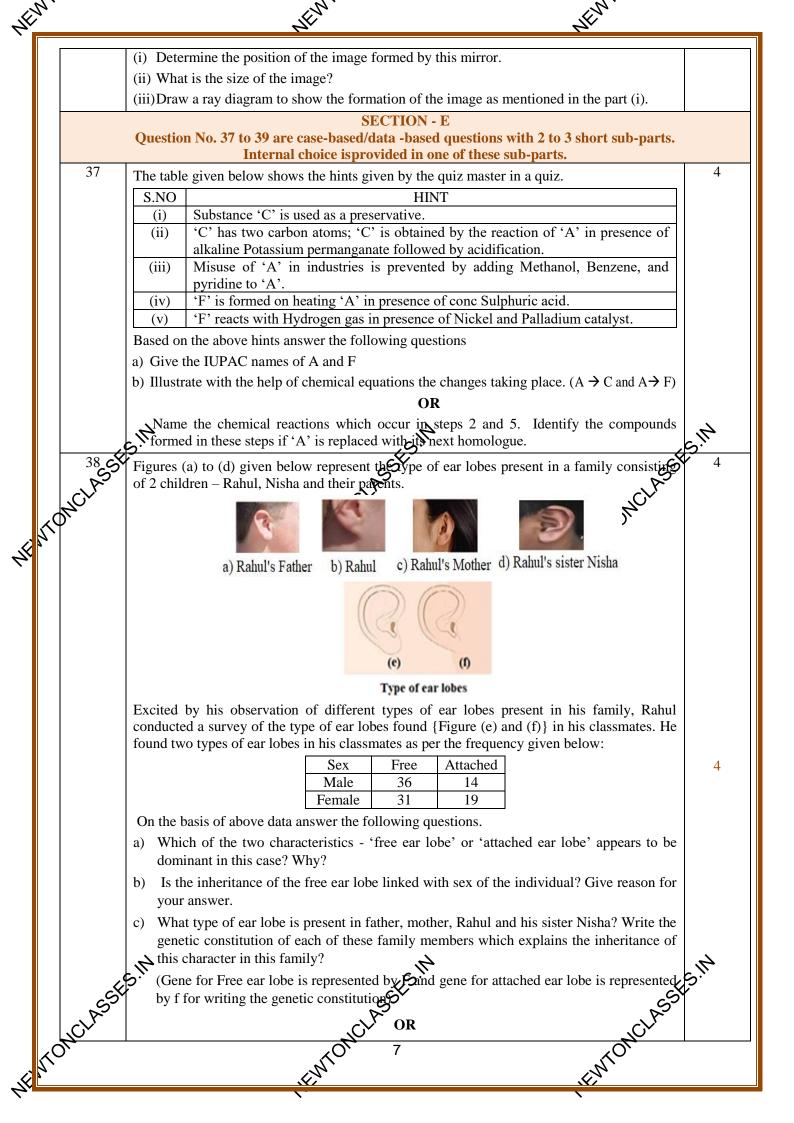
| | 4 | On adding dilute sulphuric acid to a test tube containing a metal 'X', a colourless gas is produced when a burning match stick is brought near it. Which of the following correctly represents metal 'X'? a) Sodium | | | |
|---|--|--|-----|--|--|
| | | b) Sulphur | | | |
| | | c) Copper | | | |
| | | d) Silver | | | |
| | 5 | Which one of the following correctly represents Sodium oxide? | | | |
| | | which one of the following correctly represents Sodium oxide? | | | |
| | | a) $Na^{+2} = \begin{bmatrix} x \times x \\ x \times 0 \times x \\ x \times x \times x \end{bmatrix}^{-2}$ b) $2 Na^{+} \begin{bmatrix} x \times x \\ x \times 0 \times x \\ x \times x \times x \end{bmatrix}^{-2}$ | | | |
| | | c) $2 \text{ Na}^{\dagger} 2 \begin{bmatrix} x \times x \\ x \times 0 \times x \\ x \times x \end{bmatrix}^{-1}$ c) $\text{Na} \begin{bmatrix} x \times x \\ x \times x \\ x \times x \end{bmatrix}^{-2}$ | | | |
| | 6 | An element with atomic number will form a basic oxide. | . 1 | | |
| | An element with atomic number will form a basic oxide. An element with atomic number will form a basic oxide. An element with a will form a basic oxide. An element with a will form a basic oxide. An element with a will form a basic oxide. An element with a will form a basic oxide. An element with a will form a basic oxide. An element with a wil | | | | |
| | c٤ | b) 17 (2,8,7) | 5 | | |
| | S | c) 14 (2,8,4) | | | |
| | | d) 11 (2,8,1) | | | |
| | 7 | d) 11 (2,8,1) An element 'M' has 50% of the electrons filled in the 3 rd shell as in the 2nd shell. The atomic number of 'M' is: a) 10 | | | |
| ٠ | | atomic number of 'M' is: | | | |
| | | | | | |
| | | b) 12 | | | |
| | | c) 14 | | | |
| | | d) 18 | | | |
| | 8 | Generally food is broken and absorbed within the body of organisms. In which of the | 1 | | |
| | | following organisms is it done outside the body? | | | |
| | | a) Amoeba | | | |
| | | b) Mushroom | | | |
| | | c) Paramoecium | | | |
| | | d) Lice | | | |
| | 9 | Receptors are usually located in sense organs. Gustatory receptors are present in | 1 | | |
| | | a) tongue | | | |
| | | b) nose | | | |
| | | c) eye | | | |
| | | d) ear | | | |
| | 10 | A farmer wants to grow banana plants genetically similar enough to the plants already available in his field. Which one of the following methods would you suggest for this purpose? | | | |
| | | | | | |
| | | Budding | | | |
| | 4 | a) Regeneration Budding c) Vegetative propagation d) Sexual reproduction 2 | | | |
| | | d) Sexual reproduction | | | |
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| Ċ | 12 | χO ^{N2} 2 χO ^{N2} | | | |
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| 11 | Height of a plant is regulated by: a) DNA which is directly influenced by growth hormone. b) Genes which regulate the proteins directly. c) Growth hormones under the influence of the enzymes coded by a gene. d) Growth hormones directly under the influence a gene. | 1 |
| 12 | A sportsman, after a long break of his routine exercise, suffered muscular cramps during a heavy exercise session. This happened due to: a) lack of carbon dioxide and formation of pyruvate. b) presence of oxygen and formation of ethanol. c) lack of oxygen and formation of lactic acid. | 1 |
| 13 | d) lack of oxygen and formation of carbon dioxide. An object is placed in front of a convex mirror. Its image is formed: a) at a distance equal to the object distance in front of the mirror. b) at twice the distance of the object in front of the mirror. c) half the distance of the object in front of the mirror. d) behind the mirror and it's position varies according to the object distance. | 1 |
| 14 ONCLASS | When light enters the atmosphere it strikes on extremely fine particles, which deflect the rays of light in all possible directions, This is due to - a) reflection of light b) atmospheric refraction c) scattering of light d) dispersion of light | SIM |
| 15 | In 1987, an agreement was formulated by the United Nations Environment Programme (UNEP) to freeze the production of "X" to prevent depletion of "Y". "X" and "Y" respectively referred here are: a) Ozone; CFCs b) CFCs; rays UV c) CFCs; Ozone d) UV rays; Diatomic oxygen | 1 |
| 16 | Which of the following features relates to biodegradable substances? a) Broken down by biological processes b) Remain inert c) Persist in environment for long time d) May harm the ecosystem | 1 |
| | Question No. 17 to 20 consist of two statements – Assertion (A) and Reason (R). Answer these questions selecting the appropriate option given below: a) Both A and R are true, and R is the correct explanation of A. b) Both A and R are true, and R is not the correct explanation of A. c) A is true but R is false. d) A is false but R is true. | |
| 17 18 cs/ | Assertion: Rusting of Iron is endothermic in nature. Reason: As the reaction is slow, the release of heat is barely evident. Assertion: Probability of survival of an organism produced through sexual reproduction is more than that of organism produced through asexual mode. Person: Varietions provide adventagency individuals for survival. | 1 1 5 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 |
| AK TONGY | Assertion: Probability of survival of an organism produced through sexual reproduction is more than that of organism produced through asexual mode. Reason: Variations provide advantages to individuals for survival. | |

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| • | 19 | Assertion: A compass needle is placed near a current carrying wire. The deflection of the compass needle decreases when the magnitude of the current in the wire is increased. Reason: The strength of a magnetic field at a point near the conductor increases on increasing the current. | 1 |
| | 20 | Assertion: Biodegradable substances result in the formation of compost and natural replenishment. Reason: It is due to breakdown of complex inorganic substances into simple organic substances. | 1 |
| | | Section-B | |
| | 21 | Question No. 21 to 26 are very short answer questions Dil. HCl is added to Zn granules." How will you prove that chemical change has taken place here? Support your response with two arguments. | 2 |
| | 22 | State the post-fertilisation changes that lead to fruit formation in plants. | 2 |
| | 23 | What is the purpose of making urine in the human body? Name the organs that stores and releases the urine. OR | 2 |
| | | Why do arteries have thick and elastic walls whereas veins have valves? | |
| . J. | 24 NONCLASS | The refractive indices of three media are given below: Medium | |
| | 25 | A piece of wire of resistance R is cut into three equal parts. These parts are then connected in parallel. If the equivalent resistance of this parallel combination is R_1 , what is the value of the ratio R_1 : R ? OR | 2 |
| | | Refer to the image below and state how the magnetic field pattern indicates regions where the magnetic field is stronger outside the magnet? What happens to the magnetic field when the current in the circuit is reversed? | |
| | TONOLASS | solenoid magnetic field line | W. |
| | MONO. | WIONCY 4 CINTONCY | |
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| | 26 | Study the food chain given below and answer the que | estions that follow: | 2 |
| | | Leaf ———— Caterpillar —— | Chameleon | |
| | | Mongoose | Snake | |
| | | a) If the amount of energy available at the third much energy will be available at the producer le | evel? Justify your answer. | |
| | | b) Is it possible to have 2 more trophic levels in trophic level? Justify your answer. | this food chain just before the fourth | |
| | | Section-C | | |
| | 27 | Question No. 27 to 33 are short. The given reaction shows one of the processes | | 3 |
| | | Manganese. | to extract the metals like from and | |
| | | $MnO_2(s) + Al(s) \rightarrow Mn(l) +$ | $Al_2O_3(s) + Heat$ | |
| | | a) Give reason why the above reaction is known a | | 4 |
| | ck | dr) Identify the substance oxidised and reduced in the | | S . |
| | | c) Give a reason why Aluminium is profesably use | | 3 |
| AK M | NC ASS | An element 'M' with electronic configuration 2 8 anions. Write the chemical formulae of the comporeason the nature of the bond formed by element 'conductivity of the compounds formed vary with response. | ounds formed. Predict with the uitable | 3 |
| | | OR | vas a blook commound (V) which when | |
| | | A reddish-brown metal 'X', when heated in air, giv heated in presence of H ₂ gas gives 'X' back. 'X' is this refined form of 'X' is used in electrical wiring. | s refined by the process of electrolysis; | |
| | 29 | Identify 'X' and 'Y'. Draw a well-labeled diagram to | | 3 |
| | | We are advised to take iodised salt in our diet by body. | | |
| | 30 | What is the probability of a girl or a boy being born i | • | 3 |
| | 31 | (i) Explain why the refractive index of any materia(ii) In the figure below a light ray travels from air in a reason why the ray does not deviate at the block. | nto the semi-circular plastic block. Give | 1+1+1 |
| | | plastic block 40° | normal | |
| | | light ray | | |
| | 22 | (iii) Complete the ray diagram of the above scenar plastic block from the top flat end. | | 4 |
| | 32 SE | (i) State the law that explains the heating effect of properties in an electrical circuit. (ii) List the factors on which the resistance of a conditional conditions. | current with respect to the measurable ductor depends. | 2+1 |
| | 40° | (ii) List the factors on which the resistance of a control of | W. | [|
| | , | | The | |
| 4 | | W. | <u> </u> | |
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| • | | 33 | Anannya responded to the question: Why do electrical appliances with metallic bodies are connected to the mains through a three pin plug, whereas an electric bulb can be connected with a two pin plug? She wrote: Three pin connections reduce heating of connecting wires. (i) Is her answer correct or incorrect? Justify. (ii) What is the function of a fuse in a domestic circuit? | 2+1 | |
| | | Section-D | | | |
| | | Question No. 34 to 36 are long answer questions. | | | |
| | | 34 | a) Rehmat classified the reaction between Methane and Chlorine in presence of sunlight as a substitution reaction. Support Rehmat's view with suitable justification and illustrate the reaction with the help of a balanced chemical equation. b) Chlorine gas was prepared using electrolysis of brine solution. Write the chemical equation to represent the change. Identify the other products formed in the process and | 5 | |
| | | | give one application of each. | | |
| | | | OR | | |
| | | | Raina while doing certain reactions observed that heating of substance 'X' with vinegar like smell with a substance 'Y' (which is used as an industrial solvent) in presence of conc. Sulphuric acid on a water bath gives a sweet-smelling liquid 'Z' having molecular formula C ₄ H ₈ O ₂ . When heated with caustic soda (NaOH), 'Z' gives back the sodium salt of and the compound 'Y'. | | |
| | | nClifeSt | | 5.14 | |
| | | 3350 | Given below are certain situations. Analyze and describe its possible impact on a person | 5 | |
| | , | 12° | a) Testes of a male boy are not able to descend into scrotum during his embryonic development. b) Vas deferens of a man is origged. c) Prostate and seminal varieles are not functional. | | |
| | 20 | | development. | | |
| N/K | • | | b) Vas deferens of a man is this ged. | | |
| ` | | | c) Prostate and seminar vesteres are not functionar. | | |
| | | | d) Egg is not fertilised in a human female. | | |
| | | | e) Placenta does not attach to the uterus optimally. OR | | |
| | | | a) A doctor has advised Sameer to reduce sugar intake in his diet and do regular exercise after checking his blood test reports. Which disease do you think Sameer is suffering from? Name the hormone responsible for this disease and the organ producing the hormone. | 3+2 | |
| | | | b) Which hormone is present in the areas of rapid cell division in a plant and which hormone inhibits the growth? | | |
| | | 36 | normone minoris the growth: | 1+2+2 | |
| | | | | | |
| | | | The above image shows a thin lens of focal length 5m. | | |
| | | | (i) What is the kind of lens shown in the above figure? | | |
| | | | (ii) If a real inverted image is to be formed by this lens at a distance of 7m from the optical | | |
| | | | centre, then show with calculation where should the object be placed? | | |
| | | | (iii) Draw a neatly labelled diagram of the image formation mentioned in (ii) | 4 | |
| | | MCLASSE | A 10 cm long pencil is placed 5 cm in front of a concave mirror having a radius of curvature of 40 cm. | 9 2+1+2 | |
| | | 70, | | | |
| | \$\square{\square}{\square{}} | <i>)</i> ` | of 40 cm. (EWIONCIAS (EWIONCIAS | | |
| 1/k | , | | | | |
| | | | | | |



Suresh's parents have attached earl obes. What type of ear lobe can be seen in Suresh and his sister Siya? Explain by giving the genetic composition of all. 39 1+1+2 Vinita and Ahmed demonstrated a circuit that operates the two headlights and the two sidelights of a car, in their school exhibition. Based on their demonstrated circuit, answer the following questions. (i) State what happens when switch A is connected to a) Position 2 b) Position 3 (ii) Find the potential difference across each lamp when lit. (iii) Calculate the current (iv) Show, with calculations, which type of lamp, 4.0Ω or 12Ω , has the higher power *******a) in each 12Ω lamp when lit.

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